



## Altered Product Pattern of a Squalene-Hopene Cyclase by Mutagenesis of Active Site Residues

## Thorsten Merkofer<sup>a</sup>, Catherine Pale-Grosdemange<sup>b</sup>, Karl Ulrich Wendt<sup>c</sup>, Michel Rohmer<sup>b</sup> and Karl Poralla<sup>a\*</sup>

- Mikrobiologie / Biotechnologie, Universität Tübingen, D-72076 Tübingen, Germany
   Institut Le Bel, Université Louis Pasteur / CNRS, F-67070 Strasbourg, France
- <sup>c</sup> Institut für Organische Chemie und Biochemie, Universität Freiburg, D-79104 Freiburg, Germany Received 8 January 1999; accepted 12 January 1999

Abstract. Amino acid residues lining the catalytic cavity of squalene-hopene cyclase of *Alicyclobacillus acidocaldarius* have been mutated. Alterations of His451 to Ala and Trp489 to Ala resulted in reduced enzymatic activity, while the product patterns were identical to that of the wild-type. Mutation of Phe601 to Ala led to the enhanced formation of a tetracyclic triterpene, 17-isodammara-20(21),24-diene 4, and of Tyr420 to Ala to a significant alteration of the product pattern. © 1999 Published by Elsevier Science Ltd. All rights reserved.

Keywords: Squalene cyclization, hopene, triterpene synthase

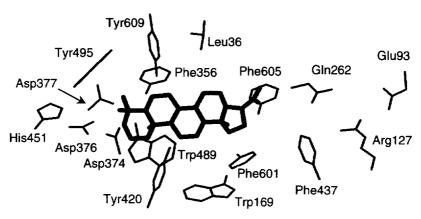
The enzymatic formation of cyclic triterpenes from squalene or (3S)-2,3-oxidosqualene is a highly complex and specific cyclization process leading to about 100 different carbon skeletons. Recently the X-ray structure of the first triterpene cyclase, namely the squalene-hopene cyclase from *Alicyclobacillus acidocaldarius*, has been elucidated, and a reaction mechanism proposed. This three-dimensional structure hopefully will form the basis for the understanding of the variability and specificity of triterpene cyclases.

The catalytic site of this membrane-bound enzyme is located in a cavity which is accessible through a non-polar channel connecting the cavity with the non-polar part of the cytoplasmic membrane where squalene 1 is dissolved. The active site comprises a cluster of aspartate residues among which Asp376 is presumably the proton donor starting the cyclization cascade (**Fig. 1**).<sup>1-4</sup> The cavity is lined mainly by aromatic residues which may expected to stabilize intermediate carbocations.<sup>2</sup> Glu45 and Glu93 abstract the proton from the final carbocation via a water molecule polarized by a hydrogen bonding network.<sup>1</sup> His451 is expected to be involved in the initial protonation of the terminal double bond of squalene by enhancing the acidity of the catalytic Asp376.<sup>1,2</sup> Tyr420 may play a structural role in the folding process of the hopene skeleton in B- or C-ring

\* Email: poralla@uni-tuebingen.de

formation. Trp489 has been proposed to stabilize the C-10 cation during A-ring formation, whereas Phe601 is in a position to stabilize the C-18 cation.<sup>2</sup>

Recently we observed in wild-type hopene cyclase some degree of unspecifity in the cyclization reaction. Along with hop-22(29)-ene 5 and diplopterol (= hopan-22-ol) 6, neohopene, tetracyclic dammaradienes and eupha-7,24-diene were detected in a yield of 0.5 - 1.5 % of the hopene-peak.<sup>5</sup>



**Figure 1.** Essential residues of the active site cavity of squalene-hopene cyclase. The model of hopene (grey) is positioned according to Reference 2.

The availability of an X-ray structure provides the opportunity to alter the active site of the cyclase by site-directed mutagenesis, and consequently to modify the product pattern. These experiments should also help to verify current ideas about the cyclization mechanism.

Site-directed mutagenesis of gene fragments was performed using the Transformer Site-Directed Mutagenesis Kit (Clontech). The mutated gene fragments were controlled by sequencing. To yield a complete cyclase gene the corresponding wild-type fragments were exchanged by the mutated ones. The cyclases were prepared from transformed *E. coli* cells by isolation of the cell membrane fraction and solubilization by Triton X-100<sup>®</sup>. This crude preparation was used for the enzymatic test. Due to the low activity of the mutated cyclases, the activity tests were performed at 50 °C for 12 h (wild-type, 1 h) as described. After extraction of squalene and the cyclization products (*n*-hexane: isopropanol, 3:2) from the assay mixture, Triton X-100<sup>®</sup> was removed by separation on a small silica gel column (*n*-hexane: ethyl acetate, 1:1). The hydrocarbon mixture (squalene und cyclization products) were analysed by GC and GC/MS.

Some properties of the mutants are described in **Table 1**. GC traces of squalene and the reaction mixture are shown in **Fig. 2**. In comparison to the wild-type, the product patterns of mutants Tyr420Ala and Phe601Ala show significant differences.

Table 1. Properties of squalene-hopene cyclase mutants	
His451Ala	reduced activity; wild-type product pattern
Tyr420Ala	reduced activity; altered product pattern
Trp489Ala	reduced activity; wild-type product pattern
Phe601Ala	reduced activity; altered product pattern

Mutant His451Ala is not fully inactivated. This result is not surprising, since the residual acidity of Asp376 may suffice for a slower protonation step. Interestingly His451 is not a conserved residue; in other hopene cyclases, Arg is at the corresponding position.

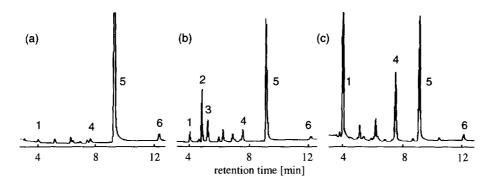


Figure 2. Gaschromatographic traces of the product pattern of wild-type squalene-hopene cyclase (a), mutant Tyr420Ala (b) and mutant Phe601Ala (c). 1, squalene; 4, 17-isodammara-20(21),24-diene; 5 hopene; 6, diplopterol. Compounds 2 - 4 are discussed in the text.

Mutant Tyr420Ala has a lower activity and yields a significantly altered distribution of products (**Fig. 2b**). According to GC/MS, compounds 2 and 3 likely correspond to a mono- and a bi- or tricyclic triterpene. These structures are under investigation and may provide insight into the function of Tyr420.

Mutant Trp489Ala resulted in a significant loss of activity without any alteration of the product pattern. The stabilization of the cation at C-10 by Trp489 seems not to be essential for the product pattern, but rather for the enhancement of the reaction velocity, which is in accordance with the proposed  $\pi$ -cation interaction.<sup>7</sup>

The product significantly enhanced in mutant Phe601Ala (no. 4 in **Fig. 2c**) is identical with one of the minor products of the wild-type enzyme, tentatively characterized as 17-isodammara-20(21),24-diene **4**. This was shown by comparison of the GC retention time and the mass spectra obtained by GC/MS. Phe601 has been proposed to be well positioned for the stabilization of cation C-18 during a 5-exo Markovnikov oriented D-ring closure. The significant increase of product **4** in mutant Phe601Ala corroborates this role since a loss of stabilization of the 5-exo C-18 cation will facilitate premature quenching of the intermediate. Based on experiments with substrate analogues, a similar 5-exo intermediate has been proposed for the C-ring closure by oxidosqualene-lanosterol cyclase.

This structure-based mutagenesis study provides for the first time light on structure-activity relationships in a triterpene cyclase and opens the field for the engineering of cyclases generating unusual triterpenes.

**Acknowledgements.** This work has been supported by the Deutsche Forschungsgemeinschaft (SFB 323, Tübingen and SFB 388, Freiburg and the Graduiertenkolleg Mikrobiologie, Tübingen). Thanks are due to Susanne Schmitz for discussion and contributions. A Forschungspreis to M. R. from the Alexander-von-Humboldt Foundation is thankfully acknowledged.

## References.

- 1. Wendt, K.U.; Poralla, K.; Schulz, G.E., Science, 1997, 277, 1811-1815.
- 2. Wendt, K.U.; Lenhart, A.; Schulz, G.E., J. Mol. Biol., in press.
- 3. Feil, C.; Süssmuth, R.; Jung, G.; Poralla, K., Eur. J. Biochem., 1996, 242, 51-55.
- 4. Corey, E.J.; Cheng, H; Baker, C.H.; Matsuda, S.P.T.; Li, D.; Song, X., J. Am. Chem. Soc., 1997, 119, 1277-1288.
- 5. Pale-Grosdemange, C.; Feil, C.; Rohmer, M.; Poralla, K., Angew. Chem. Int. Ed., 1998, 37, 2237-2240.
- 6. Deng, W.P.; Nickoloff, J.A., Anal. Biochem., 1992, 200, 81-88.
- 7. Dougherty, D., Science, 1996, 275, 1800-1804
- 8. Corey, E.J.; Virgil, S.C.; Cheng, H.; Baker, C.H.; Matsuda, S.P.T.; Singh, V.; Sarshar, S., J. Am. Chem. Soc., 1995, 117, 11819-11820.